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by

R. Todd Bronson, Jerald S. Bradshaw, Paul B. Savage, Krzysztof E. Krakowiak, and Reed M. Izatt

Department of Chemistry and Biochemistry Brigham Young University, Provo, UT 84602-4678

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Synthesis of Bis-8-Hydroxyquinoline-Armed Diazatrithia-15-Crown-5 and Diazatrithia-16-Crown-5 Ligands

R. Todd Bronson, Jerald S. Bradshaw, Paul B. Savage, Krzysztof E. Krakowiak, and Reed M. Izatt

Brigham Young University

Results and Discussion

Synthesis of Diazatrithia Crown Ethers. Secondary ring nitrogen atoms in crown ethers offer a convenient site for attachment of alkyl substituents. The crab-like synthesis of diazacrown ethers using the $bis(\alpha$ -chloroacetamide)s provides a relatively high yield method to form macrocycles containing two secondary amine functions. 1b,2 In this regard, bis(α-chloroamide) 5 was treated with various dimercaptans in MeCN using a carbonate base to form macrocylic diazatrithiadiamides 6-8 in good yields (Scheme 1). As expected, the larger 2:2 cycloaddition products, macrocyclic tetraamides 9 and 10, were also isolated in two cases in small yields. The NMR spectra of 9 and 10 were similar to those of 6 and 8, respectively. High dilution techniques helped minimize the production of these undesired by-products. Macrocyclic diazatrithia ligands 11 and 12 were prepared by reducing macrocyclic diamides 6 and 8, respectively, using a borane-THF complex. Initially, work up of the borane reduction products was done in refluxing 6 M HCl, but this process caused the formation of unexpected rearrangement and ring opened products as discussed below. Exposure to 6 M HCl at room temperature for a period of 10 minutes, along with extraction, was adequate for freeing the desired product from boron giving diazatrithia-18-crown-6 (11) and hydroxymethyl-substituted diazatrithia-15-crown-5 (12) in good yields.

The rearrangement product of 12 proved to be a new hydroxy-substituted diazatrithia-16-crown-5 (13) (Scheme 2). Ligand 13 is also of value in our research program. In an acid environment with heating, the protonated primary hydroxyl group from 12 becomes a leaving group when attacked by the neighboring ring sulfur atom. This leads to a charged epithio intermediate that is in turn attacked by water at the carbon atom most able to support a positive charge, forming 13. A minor product from this reaction resulted from the intramolecular attack by a neighboring ring nitrogen atom forming 13a in a very low-yield. A trace amount of another compound which has very similar properties to those of 13a was also observed. This material could be a result of the attack of the other ring nitrogen atom on the epithio intermediate.

Synthesis of 8-Hydroxyquinoline-substituted Ligands. Ligands 14-16 with the CHQ units attached at the CHQ 7-position were formed using Mannich reaction conditions as shown in Scheme 3. 1d,3 The best results were achieved by first forming the N,N'-bis(methoxymethyl)diazacrown ethers by stirring the diaza crowns in methanol and a slight excess of paraformaldahyde. After removal of methanol and addition of benzene to the mixtures, CHQ was added and the mixtures were refluxed. Benzene

proved to be a good reaction solvent since there were few side products. Products 14-16 were purified using radial chromatography.

Compounds 17-19 (Scheme 4) were obtained in good yields using a reductive amination procedure. ^{1a,1b} Ligands 17-19 with the 8-hydroxyquinoline side arms attached at their 2-positions were more readily isolated than compounds 14-16 with CHQ units attached at their 7-positions.

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Scheme 1. Syntheses of diazatrithiacrown ethers 11 and 12 containing two unsubstituted nitrogen atoms

Scheme 2. Rearranged crown ethers 13 and 13a

Scheme 3. Syntheses of 5-chloro-8hydroxyquinolin-7-ylmethylsubstituted crown ethers via the Mannich reaction

Scheme 4. Syntheses of 8-hydroxyquinolin-2-ylmethyl-substituted crown ethers via reductive amination